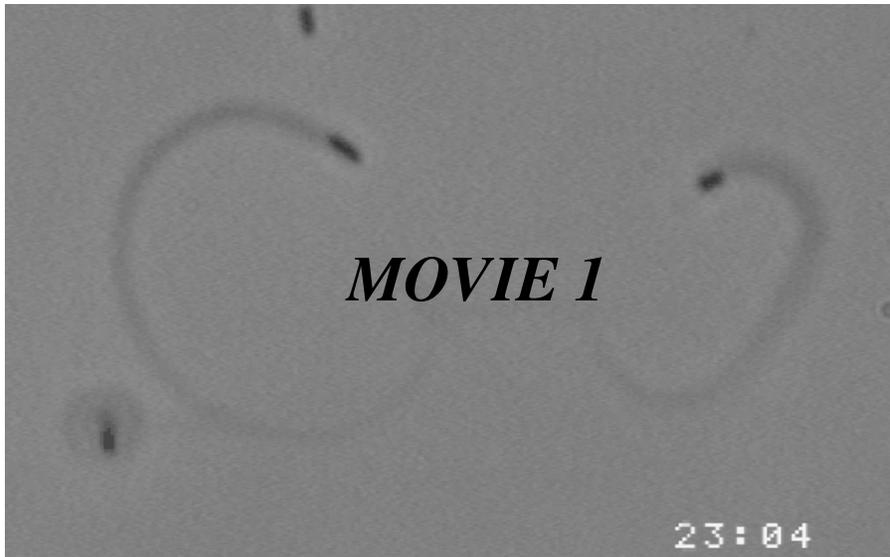


New applications of nanoscale catalysts



Catalytic conversion of chemical to mechanical energy



Example: *Listeria*

Rate $\approx 3 \mu\text{m}/\text{min}$

Movie courtesy of D. Pantaloni,
C. Le Clainche and M.-F.
Carlier, CNRS, Gif-sur-Yvette

Bacterium polymerizes a “comet tail” of actin

Provides motility inside infected cell

Requires built-in asymmetry (head different from tail)

Can we design micro- and nano-motors based on this principle?

Length scale of chemical motors

Inertial forces vs. surface tension forces

$$F = ma = \rho R^4 / \tau^2 \quad F = \gamma R$$

Surface tension dominates when $R^3 \ll \gamma \tau^2 / \rho$

What is the relevant “motor” timescale τ ?

	Speed (km/h)	Body lengths /s
Cheetah	111	25
Human	37.5	5.4
Bacteria (flagellar)	1.5×10^{-4}	10

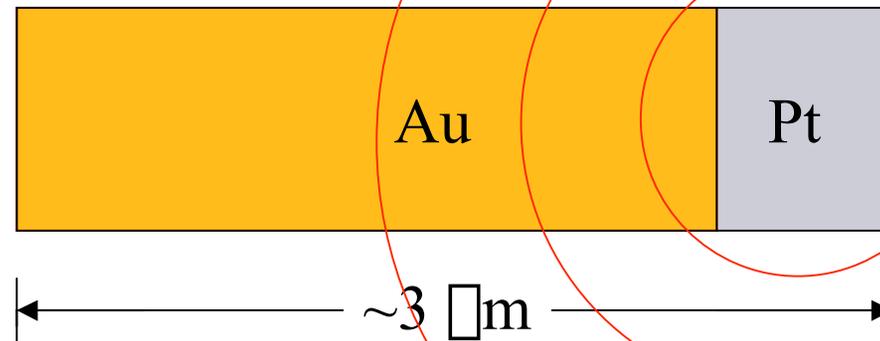
$$\tau \sim 0.1 \text{ s}$$

$$\gamma = 72 \text{ mN/m}$$

$$\rho \sim 1$$

$$R \leq 0.1 \text{ mm}$$

A simple surface tension motor: a cylinder with catalyst at one end



Pt end catalytically decomposes hydrogen peroxide:

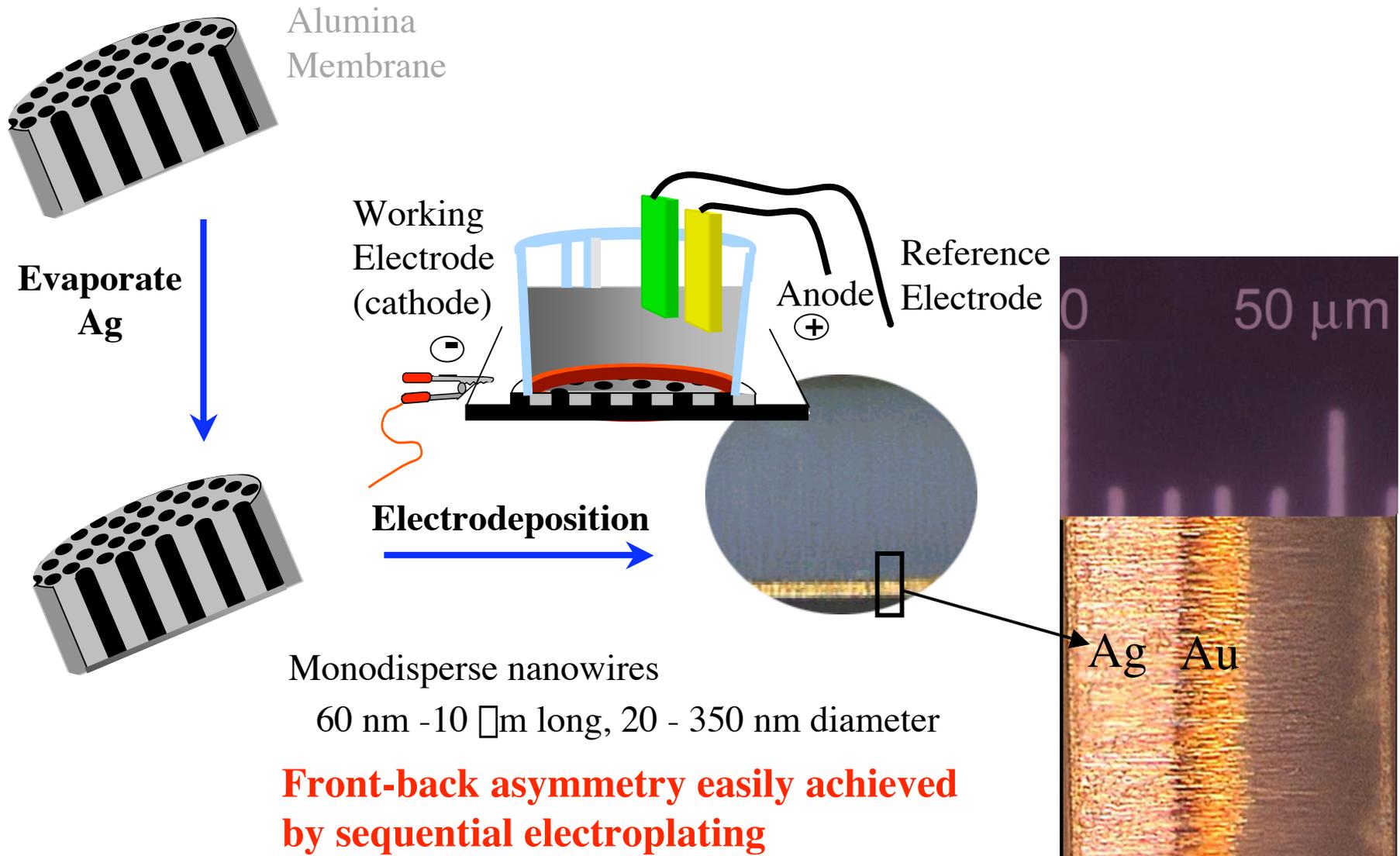


Spontaneous reaction occurs **only** at Pt tip

Experiments: Wally Paxton
 Ayusman Sen
 Tom Mallouk

Theory: Paul Lammert
 Vin Crespi
 Tom Mallouk

Electrochemical replication of porous membranes

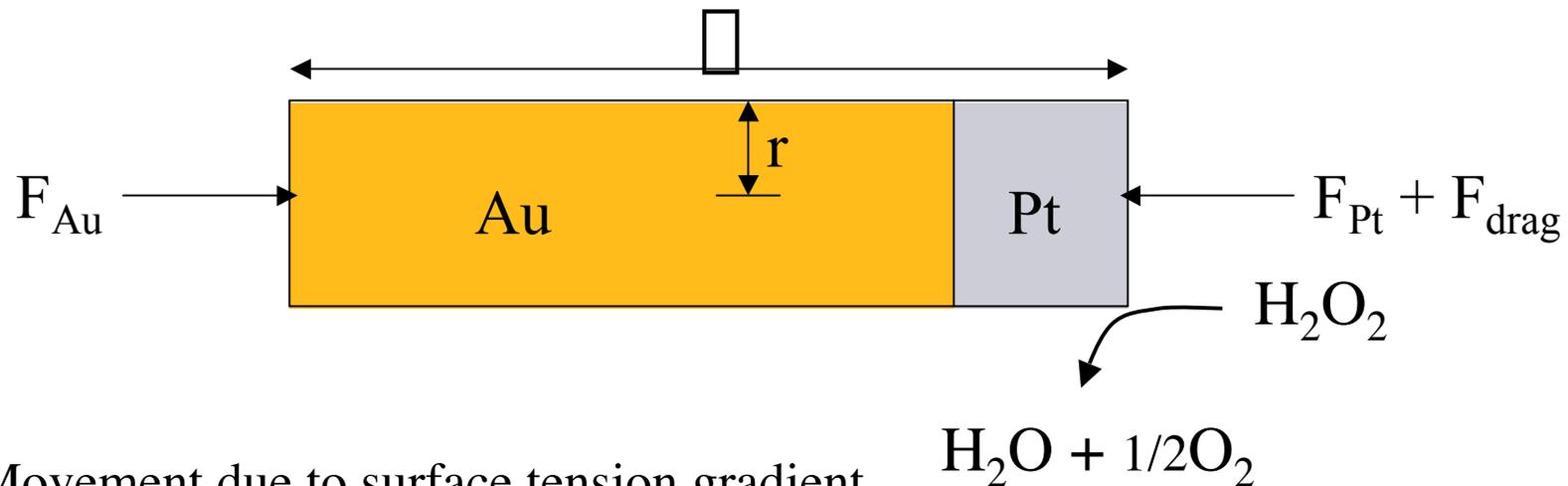


Reactive motion: Au-Pt wires in 2.5% H₂O₂



MOVIE 2

Surface tension mechanism



- Movement due to surface tension gradient
- At constant v_t , $F_{Au} + F_{Pt} + F_{drag} = 0$
- $2\pi r(\gamma_{Au} - \gamma_{Pt}) \cdot rxn\ rate \approx 6\pi\mu v_t r \rightarrow v_t \sim \gamma \sim r^2 / \mu$
- Similar method demonstrated for turning micro-gears (**Dr. Jeff Catchmark** - PSU)

Better catalysts (higher rate) needed for scaling to sub-micron regime

Assembly of nanoparticles is also important for making catalytic micro-machines

Systems based on other fuels (e.g., glucose) needed for other media (e.g., in vivo)